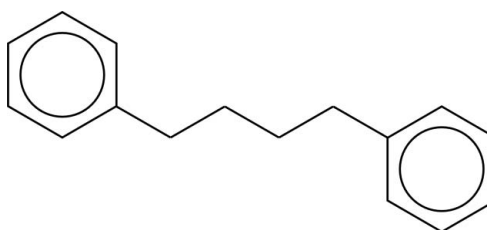


## 1,4-Diphenylbutane

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michel.fleck@univie.ac.atAll molecule of the title compound, C<sub>16</sub>H<sub>18</sub>, centrosymmetric.Received 31 October 2005  
Accepted 8 November 2005  
Online 16 November 2005

## Comment

Although a simple molecule, the crystal structure of 1,4-diphenylbutane has not hitherto been described in the literature. Bonsma & Nowacki (1959) reported the crystal structure of 2,3-diphenylbutane. In that paper, they also presented the cell parameters of 1,4-diphenylbutane, but no structural analysis had been conducted.



## Key indicators

Single-crystal X-ray study  
T = 293 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$   
R factor = 0.048  
wR factor = 0.143  
Data-to-parameter ratio = 13.3For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

1,4-Diphenylbutane, (I), was found as an impurity in the distilled raw product resulting from a reaction using lithium diisopropylamide (LDA) as reagent. As confirmed by the supplier, 1,4-diphenylbutane is an intrinsic side product of commercial LDA, arising from the production procedure which can be traced back to the experiments of Ziegler *et al.* (1934). In fact, lithiation of styrene and subsequent hydrolysis with diisopropylamine leads to LDA, ethylbenzene and, as dimerization product, 1,4-diphenylbutane.

Comparing the literature boiling point of 1,4-diphenylbutane [417–419 K at 4–5 Torr (Suga *et al.*, 1972)] and of fractions containing the desired product (428–430 K at 11 Torr), it is not surprising that separation by distillation is not successful. 1,4-Diphenylbutane was isolated from the distilled raw product by vacuum-flash chromatography over silica gel with light petroleum–ethyl acetate (20:1 *v/v*) as the eluting solvent. *n*-Hexane was used as solvent for the crystallization of suitable crystals.

The molecular geometry of 1,4-diphenylbutane does not exhibit any unusual chemical features. Aromatic C–C bond distances range from 1.365 (2) to 1.3814 (19) Å, while the aliphatic C–C bond distances are 1.5211 (17) and 1.520 (2) Å. C–H distances range from 0.946 (19) to 0.992 (17) Å (no constraints were applied in the refinement). The molecule is centrosymmetric; the mid-point of the central C–C bond is located on a centre of inversion. The butane chain is therefore in a perfect *trans* conformation [torsion angle C7–C8–C8<sup>i</sup>–C7<sup>i</sup> = 180°; symmetry code: (i) 2 – x, –y, 1 – z].

## Experimental

After aqueous work-up of an LDA reaction and concentration of the organic phase, the remainder was distilled at 428–430 K and 11 Torr. 1,4-Diphenylbutane was isolated from this crude product by vacuum-flash chromatography over silica gel with light petroleum–ethyl acetate (20:1 *v/v*) as the eluting solvent. Crystals were obtained by dissolving diphenylbutane in *n*-hexane and allowing the solvent to evaporate slowly at normal pressure. The synthesis yielded small colourless crystals up to a size of 1 mm.

### Crystal data

$C_{16}H_{18}$   
 $M_r = 210.30$   
 Monoclinic,  $P2_1/c$   
 $a = 5.787$  (1) Å  
 $b = 8.776$  (2) Å  
 $c = 12.617$  (3) Å  
 $\beta = 91.25$  (3)°  
 $V = 640.6$  (2) Å<sup>3</sup>  
 $Z = 2$

$D_x = 1.090$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 1288 reflections  
 $\theta = 26.6$ – $1288^\circ$   
 $\mu = 0.06$  mm<sup>-1</sup>  
 $T = 293$  (2) K  
 Prism, colourless  
 0.10 × 0.06 × 0.05 mm

### Data collection

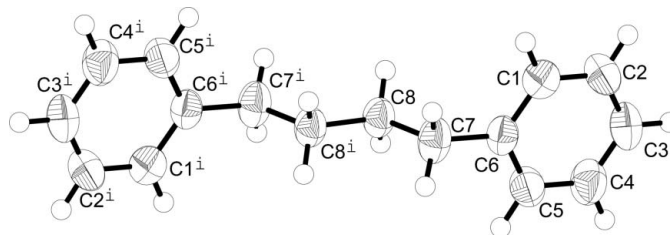
Nonius KappaCCD diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan  
 (DENZO; Otwinowski & Minor, 1997)  
 $T_{\min} = 0.994$ ,  $T_{\max} = 0.997$   
 2835 measured reflections

1464 independent reflections  
 1139 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.012$   
 $\theta_{\text{max}} = 27.5^\circ$   
 $h = -7 \rightarrow 7$   
 $k = -11 \rightarrow 11$   
 $l = -16 \rightarrow 16$

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.048$   
 $wR(F^2) = 0.143$   
 $S = 1.07$   
 1464 reflections  
 110 parameters  
 All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0749P)^2 + 0.0552P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.002$   
 $\Delta\rho_{\text{max}} = 0.11$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.12$  e Å<sup>-3</sup>  
 Extinction correction: *SHELXL97*  
 Extinction coefficient: 0.09 (3)



**Figure 1**  
 The molecular structure of (I), shown with displacement ellipsoids drawn at the 50% probability level. [Symmetry code: (i)  $2 - x, -y, 1 - z$ .]

Data collection: *COLLECT* (Nonius, 2003); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *SCALEPACK* and *DENZO* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *DIAMOND* (Version 2.1; Bergerhoff *et al.*, 1997); software used to prepare material for publication: *SHELXL97*.

The authors thank the International Centre for Diffraction Data for financial assistance of this work (grant No. 90–03 ET).

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